

Report

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PEG
$$R_2$$
-NCS R_2 -NCS R_1 R_2 -NCS R_2 -NCS R_1 R_2 -NCS R_2 -NCS R_2 -NCS R_1 R_1 R_2

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Traceless and Stereoselective Synthesis of Tetrahydro-β-carbolinethiohydantoins by Microwave Irradiation

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Combinatorial chemistry provides a powerful tool for an incessant demand of novel chemical entities from medicinal chemistry and has greatly influenced the whole gamut of drug discovery process. In recent years it has been experiencing a paradigm shift with special emphasis laid on liquidphase methodologies.^{1,2} Though the idea of a macromolecule acting as a support for organic reactions was born with the synthesis of peptides,³ it has now grown into an independent methodology for creating a variety of nonpeptidic, drug-like molecular libraries. 4,5 signaling new vistas in classical and combinatorial synthesis.6 In an inimitable way the soluble polymer supported organic synthesis retains the advantage of the homogeneity of the classical organic reaction whose progress can be monitored by conventional analytical techniques.⁷ In addition to this, the easy product isolation by precipitation, high yields, and compatibility with microwave (MW) irradiation^{8,9} have made this method highly suitable for a multistep organic synthesis involving a variety of functional group interconversions. Polymer supported reactions in which the macromolecular carrier does not leave behind any structural imprint on the target molecules are referred to as traceless syntheses, which are a new trend in both solid-phase¹⁰ and liquid-phase¹¹ methodologies.

Heterocycles are versatile molecular scaffolds, on to which meaningful structural diversity can be introduced to vary the stereochemical and physicochemical properties of the resulting compounds, thus tailoring them for better membrane transport and receptor binding in biological systems. Tetrahydro- β -carbolines are one such class of semirigidized tricyclic heterocycles, 12 which are found to occur as mammalian¹³ and plant alkaloids, like jafraine, ¹⁴ exhibiting a wide range of biological activities. ^{15–17} The tetrahydro- β -carbolines such as demethoxyfumitremorgins have been isolated from the fungal species, some of which have been found to exhibit antiviral, 18 topoisomerase II inhibiting, 19 and protein kinase and cell cycle inhibiting²⁰ activities (Figure 1). Because of their potential biological properties, these tetra- and polycyclic β -carbolines have been synthesized by classical^{21,22} as well as solid-phase²³⁻²⁶ methods.

The design concept of the presently synthesized molecular library has originated from the recognition of the biological role of the thiohydantoin moiety. A large number of thiohydantoins have been associated with calcium mobilization,²⁷ inhibition of glycogen phosphorylases,²⁸ aldose reductase inhibition,²⁹ anti-HIV,³⁰ and anti-leukemic³¹ activities. Thus the generation of a combined tetracyclic skeleton ressembling drug-like molecules has a substantial intellectual appeal. The present paper demonstrates successful application of liquid-phase methodologies toward the parallel stereoselective synthesis of thiohydantoin fused tetrahydro- β -carbolines powered by the microwave irradiation. Application of microwave irradiation for organic reactions has emerged as a practical technique to reduce reaction times and enhance the yields, thus speeding up the drug discovery process. 32,33 It has been applied by our group, leading to the generation of the molecular libraries of thioxotetrahydro pyrimidinones³⁴ and hydantoins.35

The enhanced loading capacity of the dihydric poly-(ethylene glycol), PEG-4000(1), has been used in the stepwise synthesis toward the target molecular libraries (Scheme 1). The synthesis was initiated by the generation

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Figure 1. Naturally occurring tetrahydro- β -carbolines.

Scheme 1. Synthetic Route to Tetracyclic- β -carbolinethiohydantoins 5

of the PEG immobilized bis-ester **3** from the commerciably available Boc protected L-tryptophan **2**. This esterification reaction was brought about via the dicyclohexyl carbodiimide (DCC) activated ester in the presence of catalytic quantities of (dimethylamino)pyridine (DMAP), in a microwave oven under open vessel conditions. The power and time variation studies in this step have lead to an optimized experimental condition of 100 W power and 20 min as the reaction time to obtain maximum yields. The reaction time under refluxing conditions at the same temperature was 48 h to reach completion and has been brought down to 20 min, which shows the remarkable power of microwave irradiation.

Construction of the β -carboline skeleton was perceived in terms of a [5+1] approach for the six-membered heterocycles, which needed the generation of a nucleophilic amino group. The preferential cleavage of the amide bond (N-CO) over the acyl-oxygen bond (O-CO) in the polymer ester conjugate 3 was achieved in microwave cavity with 30% TFA in dichloromethane. The initial study to remove the Boc group was ensured by the disappearance of the crude NMR signal of tert-butyl group at 1.4 ppm. The in situ generated amine was treated with an appropriate aldehyde to obtain an imine which underwent an acid-catalyzed intramolecular cyclization. 36,37 Identical results were observed with the simultaneous addition of TFA and the aldehydes, which resulted in the sequential deprotection—nucleophilic addition and cyclization. The introduction of the first point of structural diversity was thus accomplished in one pot.

The transient imine was found to undergo an intramolecular electrophilic ring closure at the C-2 position of the indole nucleus, forming a tricyclic polymer intermediate 4. This cyclization step which required 24 h under refluxing conditions in chloroform also proceeded remarkably well in 20 min under microwave conditions. Cyclocondensation of aldehydes with PEG bound tryptophan 3 resulted in the formation of cis and trans diastereomers around 1:1 ratio. Building up of the terminal thiohydantoin ring across the N-2/C-3 bond of the tetrahydro- β -carboline skeleton was achieved by the reaction of β -carboline derivatives 4 with various isothiocyanates. This nucleophilic addition was achieved without the use of any metal salt or activating agent. The application of MW (200 W) conditions in this step also resulted in fast completion of intramolecular cyclative cleavage in 30 min in ethylene dichloride. The reactive thioureas would induce an acyl-oxygen bond cleavage of the ester by a S_Ni reaction via N-CO bond concomitant formation leading to a traceless and stereoselective synthesis of substituted indole alkaloids 5.

The stereochemical outcome in the Pictet—Spengler reaction has been of immense interest, since the diastereomeric proportion depends on the acidity of the medium, steric bulk of the ester function, and the substituent on the nitrogen of the tryptophan moiety.³⁸ We have observed that the ester polymer conjugates also lead to the formation of a mixture of *cis* and *trans* diastereomer which varies with the nature of the carbonyl group in *p*-toluenesulfonic acid (*p*-TSA)

Figure 2. ORTEP diagram of **5a** ($R = R_1 = CH_3 - (CH_2)_3 - CH_2 -$

Table 1. Synthesis of Tetracyclic Tetrahydro-β-carbolines

Table 1. Synthesis of Tetracyclic Tetranydro-p-carbolines					
Entry	R₁CHO	R₂NCS	crude yield ^a (%)	crude purity ^b	(%) LR-mass
5а	н	∕ \\ NCS	91	84	370(M +1)
5b	н	NCS	98	97	404(M+1)
5c	н	=/ ^{NCS}	93	92	353
5d	н	NCS	85	90	381
5e	H	NCS	85	84	397(M+1)
5f	H	NCS	94	73	431
5g	H	∕ NCS	91	92	418(M+1)
5h	H	NCS	98	95	452(M+1)
5i	H	NCS	96	98	429
5j	H	=/-NCS	90	97	402
5k	н	NCS	92	83	437
51	H	F—NC	S 96	80	407
5m	H	∕ NCS	84	87	433
5n	H 0	CINCS	91	93	451
50	$H^{\bigcup_{i=1}^{n}}$	CI—NO	CS 85	90	451

^a Yields were based on the weight of the isolated crude samples. ^b Purity was estimated by HPLC without considering the peak of excess R-NCS.

catalyzed Pictet-Spengler reaction at room temperature.³⁹ The MW conditions employed during the present work favor the thermodynamically stable trans congeners than the cis isomers. 40-42 Further, in the formation of the thiohydantoin ring D, the orientation of the nitrogen lone pair also requires the bulkier group to be in trans orientation to keep the electronic repulsions to the minimum. The unprecedented 100% trans diastereomeric selectivity under the MW conditions has been confirmed by ¹³C NMR and homo- and heteroCOSY experiments. HPLC analysis of crude products also indicated only one diastereomer was obtained (Table 1).

Additional support for the configuration of the structure as trans diastereomer was obtained from the X-ray diffraction studies which indicated R,S configuration at the two chiral centers depicted in the ORTEP diagram (Figure 2). From the ORTEP diagram for 5a, it is also clear that the rings C and D are trans fused and are nonplanar. The hydrogens on C-3 and C-13 are anti-periplanar, and the substituent at C-13 is α-oriented.

In summary, we have demonstrated a microwave-assisted traceless, liquid-phase methodology⁴³ to assemble substituted indole alkaloids with exceedingly high stereoselectivity. All the steps in this synthetic sequence have been accomplished under focused microwave irradiation, resulting in significantly reduced reaction times-from hours to minutes-and enhanced yields. It should be noted that polymer supported intermediates and polymer itself are stable during the harsh MW irradiation. The presently reported rapid synthesis of tetracyclic tetrahydro- β -carboline pharmacophores with two points of diversity has the potential for the creation of a diverse array of polycyclic fused heterocyclic systems, which closely resemble biologically active natural products.

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Supporting Information Available. Representative experimental procedures, spectral data of compounds 5a-5o, and the bond angle and bond length data for 5a. Comparison of microwave and thermal conditions for the above reactions are also included. This material is available free of charge via the Internet at http://pubs.acs.org.

References and Notes

- (1) Gravert, D. J.; Janda, K. D. Chem. Rev. 1997, 97, 489-509.
- (a) Sun, C. M. Comb. Chem. High Throughput Screening **1999**, 2, 299–309. (b) Sun, C. M. Soluble Polymer-Supported Synthesis of Heterocyclic Libraries. In Combinatorial Chemistry Methods and Protocols; Bellavance, L., Ed.; Methods in Molecular Biology Series; The Humana Press Inc.: New Jersey, 2002; Chapter 10, 345-371.

- (3) Bayer, E.; Mutter, M. Nature 1972, 237, 512-513.
- (4) Wentworth, P. Trends Biotechnol. 1999, 17, 448-452.
- (5) Lee, M. J.; Sun, C. M. Chin. Pharm. J. 2003, 55, 405-452.
- (6) Harwig, C. W.; Gravert, D. J.; Janda, K. D. Chemtracts 1999, 12, 1–27.
- (7) Yeh, C. M.; Tung, C. L.; Sun C. M. J. Comb. Chem. 2000, 2, 341–348.
- (8) Su, Y. S.; Lin, M. J.; Sun, C. M. Tetrahedron Lett. 2005, 46, 177–180.
- (9) Lin, M. J.; Sun, C. M. Synlett 2004, 663-666.
- (10) Blaney, P.; Grigg, R.; Sridharan, V. Chem. Rev. 2002, 102, 2607–2624.
- (11) Zong, Y. X.; Wang, J. K.; Yue, G. R.; Feng, L.; Song, Z. E.; Song, H.; Han, Y. Q. Tetrahedron Lett. 2005, 46, 5139– 5141.
- (12) Gremmen, C.; Brigitte, A.; Burn, M. J.; Koomen, G. J. Tetrahedron Lett. 1998, 39, 1441–1445.
- (13) Schouten, M. J.; Bruinvels, J. Anal. Biochem. 1985, 147, 401–409.
- (14) Fazai, S.; Naz, A. Tetrahedron Lett. 2002, 58, 6185-6197.
- (15) Maw, G. N.; Allerton, C. M. N.; Gbekor, E.; Million, W. A. Bioorg. Med. Chem. Lett. 2003, 13, 1425–1428.
- (16) Abdel-Moty, S. G.; Sakai, S.; Aini, N.; Takayama, H.; Kitajima, M.; El-Shorbagi, A.; Ahmed, A. N.; Omar, M. N. Eur. J. Med. Chem. 1998, 32, 1009-1017.
- (17) Allen, M. S.; Tan, Y. C.; Trudel, M. L.; Narayan, K.; Schindler, L. R.; Martin, M. J.; Schultz, C.; Hagen, T. R.; Koehler, K. F. J. Med. Chem. 1990, 33, 2343–2357.
- (18) van Maarseveen, J. H.; Scheeren, H. W.; De Clercq, E.; Balzarini, J.; Kruse, C. G. *Biorg. Med. Chem.* **1997**, *5*, 955–970
- (19) Deveau, A. M.; Labroli, M. C.; Dieckhans, C. M.; Barthen, M. T.; Smith, K. S. *Bioorg. Med. Chem. Lett.* **2001**, *11*, 1251–1255.
- (20) Osada, H.; Cui, C. B.; Onose, R.; Hanaka, F. Bioorg. Med. Chem. 1997, 5, 193–203.
- (21) O'Malley, G. J.; Cava, M. P. Tetrahedron Lett. 1987, 28, 1131–1134.
- (22) (a) Wang, H.; Ganesan, A. Tetrahedron Lett. 1997, 38, 4327–4328. (b) Wang, H.; Usui, T.; Osada, H. Ganesan, A. J. Med. Chem. 2000, 43, 1577–1585. (c) Revell, J. D.; Srinivasan, N.; Ganesan, A. Synlett 2004, 8, 1428–1430.
- (23) van Loevezijn, A.; van Maarseveen, J. H.; Stegman, K.; Visser, G. M.; Koomen, G. J. *Tetrahedron Lett.* **1998**, *39*, 4737–4740.
- (24) Fanatauzzi, P. P.; Yager, K. M. Tetrahedron Lett. 1998, 39, 1291–1294.
- (25) Wang, H.; Ganesan, A. Org. Lett. 1999, 1, 1647-1649.
- (26) (a) Klein, G. J.; Ostresh, J. M.; Nefzi, A. Tetrahedron Lett. 2003, 44, 2211–2215. (b) Yen, Y. H.; Chu, Y. H. Tetrahedron Lett. 2004, 45, 8137–8140.
- (27) Incesu, Z.; Benkli, K.; Akaalin, G. Kaplancikli, Z. A. Cell Biol. Int. 2004, 28, 267–272.
- (28) Oikonomakos, N. G.; Skamnaki, V. T.; Osz, E.; Szilagyi, L.; Somsak. L.; Docsa, T.; Toth, B.; Gergly, P. *Bioorg. Med. Chem.* 2002, 10, 261–268.
- (29) Buyukbingol, E.; Suzen, S.; Klopman, G. Farmaco 1994, 49, 443–447.

- (30) Suzen, S.; Buyukbingol, E. Farmaco 1998, 53, 525-527.
- (31) Suzen, S.; Buyukbingol, E. Farmaco 2000, 55, 246-248.
- (32) (a) Santagada, V.; Perissuti, E.; Caliendo, G. Curr. Med. Chem. 2002, 9, 1251–1283. (b) Al-Obeidi, F.; Austin. R. E.; Okonya, J. F.; Bond, D. R. S. Mini-Rev. Med. Chem. 2003, 3, 459–470. (c) Blackwell, H. E. Org. Biomol. Chem. 2003, 1, 1251–1255. (d) Swamy, K. M. K.; Yeh, W. B.; Lin, M. J.; Sun, C. M. Curr. Med. Chem. 2003, 10, 2403–2424. (e) Kappe, C. O. Angew. Chem., Int. Ed. 2004, 43, 6250–6284.
- (33) Yoon, D. S.; Han, Y.; Stark. T. M.; Haber, J. C.; Gregg, B. T.; Stankovich, S. B. Org. Lett. 2004, 6, 4775– 4778
- (34) Yeh, W. B.; Lin, M. J.; Lee, M. J.; Sun, C. M. Mol. Diversity **2003**, 7, 185–198.
- (35) (a) Lin, M. J.; Sun, C. M. Tetrahedron Lett. 2003, 44, 8739–8742. (b) Lee, M. J.; Sun, C. M. Tetrahedron Lett. 2004, 45, 437–440.
- (36) Mayer, J. P.; Davis, D. B.; Zhang, J.; Beaton, G.; Bjergarde. K.; Anderson, C. M.; Goodman, B. A.; Herrera, C. J. Tetrahedron Lett. 1996, 37, 5633-5636.
- (37) Connors, R. V.; Zhang, A. J.; Shuttleworth, S. J. *Tetrahedron Lett.* **2002**, *43*, 6661–6663.
- (38) Cox, E. D.; Hamaker, L. K.; Li, J.; Peng, Yu.; Czerwinski, K. M.; Deng, Li.; Bennet, D. W.; Cook, J. M. J. Org. Chem. 1997, 62, 44-61.
- (39) (a) Yeh, W. B.; Lin, M. J.; Sun, C. M. Tetrahedron Lett. 2003, 44, 4923–4926. (b) Campiglia, P.; Gomez-Monterrey, I.; Lama, T.; Novellino, E.; Grieco, P. Mol. Diversity 2004, 8, 427–430.
- (40) Kowalski, P.; Bojarski, A. J.; Mokrosz, J. L. Tetrahedron 1995, 51, 2737–2742.
- (41) Gremmen, C.; Willemse, B.; Wanner, M. J.; Koomen, G. J. Org. Lett. 2000, 2, 1955–1958.
- (42) Lopez-Rodriguez, M. L.; Morcillo, M. J.; Garrido, M.; Benhamu, B.; Perez, V.; de la Campa, J. G. J. Org. Chem. 1994, 59, 1583–1585.
- (43) All the microwave assisted polymer supported reactions described here were performed in a 50 mL round-bottom flask (attached to the reflux condenser) with the CEM Discover Microwave System at a frequency of 2450 Hz (0-300 W). The spectral data for **5a** is as follows: $[\alpha]_D^{20} = -46.41^{\circ}$ (c = 1.03; CH₂Cl₂); ¹H NMR (300 MHz, CDCl₃) δ 8.01 (s, 1H, N_3 -H), 7.46 (d, 1H, J = 7.8 Hz, C_8 -H), 7.35 (d, 1H, $J = 7.8 \text{ Hz}, \text{ C}_9 - \text{H}), 7.24 \sim 7.12 \text{ (m, 2H)}, 5.87 \text{ (t, 1H, } J =$ 6.9 Hz, C_{13} -H), 4.38 (dd, 1H, J = 5.7, 10.8 Hz, C_3 -H), 3.90 (t, 2H, J = 7.8 Hz), 3.38 (dd, 1H, J = 6.0, 15.0 Hz, C_4-H), 2.70 (dd, 1H, J=11.1, 15.3 Hz, C_4-H), 2.19 (m, 1H, C_{18} -H), 1.90 (m, 1H, C_{18} -H), 1.76 \sim 1.67 (m, 4H), $1.47 \sim 1.31$ (m, 4H), 0.99 (t, 3H, J = 7.2 Hz, CH₃), 0.87 (t, 3H, J = 4.8 Hz, CH₃); ¹³C NMR (75 MHz, CDCl₃) δ $181.3(C_1)$, $173.6(C_2)$, $136.4(C_{12})$, $132.4(C_{11})$, $126.1(C_6)$, $122.7(C_8)$, $120.2(C_9)$, $118.2(C_7)$, $111.1(C_8)$, $105.9(C_5)$, $56.8(C_3)$, $52.6(C_{13})$, $41.4(C_{14})$, $34.7(C_{18})$, $29.9(C_{19})$, $27.6(C_{15})$, $23.7(C_4)$, $22.9(C_{20}), 20.1(C_{16}), 14.0(C_{21}), 13.8(C_{17}); LRMS (FAB)$ found 370; HRMS(EI) calcd 369.1875, found 369.5316; IR (KBr, cm⁻¹) 3390, 2950, 1730, 1645, 1462.

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